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~~Solutions, Suspensions, and~~

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appear blue? | #aumsum #kids  
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#children 3 kinds of mixture  
( solution, suspension,  
colloid) Simple Distillation  
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## Colloids

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5.4 *types of colloidal  
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Colloid and Surface  
Chemistry, 2nd ed. Marcel

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**Solution**, New York, 1986, 815 p. This is an advanced book that is a very good starting point for serious theoretical considerations of the title topic.

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Hiemenz, P. (Ed.),  
Rajagopalan, R. (Ed.).  
(1997). Principles of  
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Chemistry, Revised and  
Expanded. Boca Raton: CRC

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**Solution**, <https://doi.org/10.1201/9781315274287>. COPY. This work aims to familiarize students with the fundamentals of colloid and surface science, from various types of colloids and colloidal phenomena, and

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Solution, the molecules or ions are invisible, and they do not settle out on standing. A group of mixtures called colloids (or colloidal dispersions) exhibit properties intermediate between those

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**Solution** of suspensions and solutions  
. The particles in a colloid  
are larger than most simple  
molecules; however,  
colloidal particles are  
small enough that they do  
not settle out upon  
standing.

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## **11.5 Colloids - Chemistry**

There are two parts to every colloid mixture: the particles and the dispersing medium. The colloid particles are solids or liquids that are suspended

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**Solution** in the medium. These particles are larger than molecules, distinguishing a colloid from a solution. However, the particles in a colloid are smaller than those found in a suspension. In smoke, for examples,

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**Solution**  
Solid particles from  
combustion are suspended in  
a gas.

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## **Colloid Chemistry Hiemenz Solution Bsbltd**

In chemistry, a colloid is a phase separated mixture in

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**Solution** which one substance of microscopically dispersed insoluble or soluble particles is suspended throughout another substance. Sometimes the dispersed substance alone is called the colloid; the term

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**Solution** colloidal suspension refers unambiguously to the overall mixture. Unlike a solution, whose solute and solvent constitute only one phase, a colloid has a dispersed phase and a continuous phase that arise by phase

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Solution. Typically,  
colloids do no

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