

Chemical Equilibria In Soils

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1.1 Dynamic Equilibria in Soils, 1.2 Intensity and Capacity Factors, 1.3 Elemental Composition of Soils, References, Problems, 2 Methods of Handling Chemical Equilibria 1 2.1 Equilibrium Constants, i 2,2 Concentration Versus Activity Constants, 2.3 Ionic Strength, 2.4 Activity Coefficients, 2,5 Debye-Hückel Equations, ,

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Author (s) : Lindsay, W. L. Author Affiliation : Colorado State University, Fort Collins, USA. Book : Chemical equilibria in soils. pp.449pp. Abstract : This book has been written from the viewpoint that most reactions taking place in soils can be understood and predicted from basic chemical relationships.

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This chapter describes a model for chemical equilibria in soils and solutions (the CHESS model) that can be used to assess reaction formulation in environmental chemistry research. The approach used in the CHESS model is similar to models such as MINEQL that are based on the tableau notation for describing chemical systems.

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Emphasis is placed on minerals and solid phases in soils that dissolve and precipitate, and in doing so, control the composition of the soil solution. This is the only comprehensive treatment of chemical equilibria in the soil environment that is available. Other texts deal well with the kinetics of reactions in soils but touch lightly on the equilibrium concepts.

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The author, a renowned authority in the field, focuses on reactions in the soil solution and considers aqueous phase properties. He also cogently draws distinctions between thermodynamic chemical species and kinetic chemical species.

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